

In vitro antioxidant properties of edible marine algae *Sargassum swartzii*, *Ulva fasciata* and *Chaetomorpha antennina* of Kerala coast

Oxidative stress is reported to be the main cause of various life style diseases. This stress is due to the imbalance of oxidants and anti oxidants. Seaweeds form the richest renewable source of ample bioactive compounds that promote them as a vital candidate in therapeutics, especially for drug development. Among the seaweed born macromolecules, the sulfated polysaccharides from brown, green and red algae were reported to have antioxidant activities, despite of their structural and nutritional significance. In the current study, the sulfated polysaccharides were isolated from three marine algae, *Sargassum swartzii*, *Ulva fasciata* and *Chaetomorpha antennina*. The carbohydrate content in the three edible algae was found to be 12.9%, 12.3% and 12.7% respectively. Of the three, *Sargassum swartzii* contains high sulfate content, which might be responsible for its better bioactivity. The antioxidant capability of the polysaccharides was determined by using standard *in vitro* assays and was found to be good in scavenging hydrogen peroxide. However, the total antioxidant activity and reducing power were less effective. Among the three, *Sargassum swartzii* showed superior activity and can be a good candidate for antioxidant therapies.

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Introduction

Oxidative stress induced by the free radicals has been gained a vital importance as it forms the root cause of about 200 human diseases [1]. Free radicals are highly reactive molecules with unpaired electrons and are produced during various cellular processes [2]. They represent an essential part of metabolism and aerobic life. Many of the reactive oxygen species (ROS) and reactive nitrogen species (RNS) are grouped under free radicals. These include superoxide anions (O_2^-), hydroxyl radical ($\cdot OH$), singlet oxygen, hydrogen peroxide (H_2O_2), ferric ion, nitric oxide (NO) and so on [3]. ROS and RNS are produced from both endogenous (inflammation, mental stress and cancer) and exogenous (pollutants, drugs and radiations) sources [4]. Under normal conditions ROS participate in many physiological functions such as protecting body from invading pathogens, regulates calcium concentration and act as second messengers [5]. They participate in the signal transduction of cytokine and tyrosine receptor, serine/threonine kinases and G protein-coupled receptors [3]. They are also essential in development and differentiation process [6].

Several degenerative changes in the cells and tissues due to oxidative stress can lead to many deadly diseases [7]. The oxidative stress damages nucleic acids, lipids and proteins in our body, alters cellular functions and finally results in apoptosis or necrosis [5]. Also it is responsible for the progression of diseases like diabetics, rheumatoid arthritis, myocardial infarction, cancer, post-ischemic perfusion injury, autoimmune pathologies, cardiovascular, neurodegenerative and inflammatory diseases [8]. Antioxidants can grant protection from oxidative damages and prevent the onset of many chronic diseases [9]. They are naturally present in our body (endogenous) and the additional supplementation can be done through the diet (exogenous). Natural antioxidants like ascorbic acid (vitamin C), α -tocopherol, and carotenoids are readily absorbed through diet [10]. Butyl hydroxyanisole (BHA) and butyl hydroxytoluene (BHT) are synthetic antioxidants but are proven to cause major side effects such as cancer [11]. The current trends in the pharmaceutical industries are the exploration of natural antioxidants to resolve such issues. Plants have been widely investigated as the potent source of

antioxidants. The dietary antioxidants such as α -tocopherol, ascorbic acid, carotenoids, amino acids, peptides, proteins, flavonoids and other phenolic compounds were proven effective in boosting antioxidant mechanism. The role of naturally occurring peptides in the biological system is well-established [12]. Peptides capable of developing synthetic vaccines against diseases are reported [13].

The marine world is also a rich source of bioactive molecules [14]. Among the marine organisms, seaweeds are well-explored for various bioactive compounds such as secondary metabolites, dietary fiber, minerals, lipids, proteins, omega-3 fatty acids, essential amino acids, polysaccharides and vitamins [15]. These compounds impart numerous bioactivities such as anti-oxidative, anti-inflammatory, antimicrobial and anti-cancer potential to these algae [16]. Among different algal derived compounds, sulfated polysaccharides (SPS) gained a crucial attention by the pharmaceutical industries. They are complex group of molecules with varying structure and properties. SPS are present in brown, green and red marine algae [17]. The development of a standardized algal polysaccharide based product is challenging, as their structure and properties are influenced by seasonal and climatic variations. Their high molecular weight and low bioavailability also form hurdles [18].

Despite of these difficulties, scientists have succeeded in elucidating several biological significances for algal derived sulfated polysaccharides. SPS have been hailed for the antioxidant, antitumor, immunomodulatory, anti-inflammation, anticoagulant and antimicrobial activities [18]. Fucoidan, a sulfated polysaccharide from *Undaria pinnatifida* was proven effective against hypersensitivity reactions by reducing the concentrations of both IL-4 and IL-13. Anti coagulant activity is the most studied property of marine sulfated polysaccharides and is similar to that of natural antioxidant heparin [19]. Cytotoxicity of marine sulfated polysaccharides against various cancer cell lines such as HeLa, HepG2, MCF-7 and melanoma B16 were also well proven [21]. Studies show that, the sulfate content and the molecular weight play key roles in the bioactivities of these sulfated polysaccharides [17].

Sargassum swartzii (brown), *Ulva fasciata* and *Chaetomorpha antennina* (green) are the three main macro algae found in South Kerala coast. *Sargassum swartzii* was exploited for its larvicidal

[22], anti HIV-1, anti-inflammatory and analgesic activities [23]. Anti-microbial, haemolytic [24] and anti-cancer [25] potentials of *Ulva fasciata* are also well studied. *Chaetomorpha antennina* is proven to possess anti-bacterial and antioxidant activity [26]. In the present study, we aimed to compare the antioxidant potentials of SPS from these three sources. We also have thrown light to correlate the chemical composition and the antioxidant potential among the three SPS.

Materials and Methods

Chemicals

All chemicals used in this study were of analytical grade obtained from Sisco Research Laboratories (SRL), Mumbai, Sigma-Aldrich, New Delhi.

Collection of seaweeds

Samples of *Sargassum swartzii*, *Ulva fasciata* and *Chaetomorpha antennina* were collected from Vizhinjam coast of Kerala, (Lat. 80 22' N; Long. 760 59' E on the west coast of India) during the month of January-February.

Extraction and isolation of crude sulfated polysaccharides

The collected seaweeds were washed in tap water, dried under shade, powdered and stored in airtight containers. Crude sulfated polysaccharides were isolated from all the samples using cold acidic extraction method [21]. The samples were decolorized and defatted by soaking and continuous stirring in acetone: methanol solvent mixture (7:3) and then stirred in 1N HCl for two days and then filtered. These steps were repeated twice and the filtrates were pooled and stored at 40C overnight. Then the polysaccharides were precipitated using absolute ethanol and lyophilized to obtain crude sample.

Determination of chemical composition.

The compositions of sulfated polysaccharide moieties such as total carbohydrates, sulfate, uronic acid, sulfated polysaccharide, fucose and xylose present in all extracts were determined.

Total carbohydrate content was estimated by the phenol-sulfuric acid method as described by Pham Duc Thinh et al., [27]. 5% phenol and concentrated sulfuric acid were added to the test sample (10mg/ml), incubated for 20 min and optical density (OD) was read at 490 nm. Dextrose was used as standard.

The sulfate content was determined by barium chloride-gelatin method using potassium sulfate as standard [28]. The reaction mixture was prepared by adding 2g barium chloride to

a solution of 0.6g gelatin in 200ml water which was kept overnight at 40°C. To the test solution (10mg/ml) 4% TCA and 1ml chloride – gelatin solution was added and optical density was read at 360 nm after 15 min incubation.

Uronic acid content of the extract was estimated by carbazole method using glucuronic acid as standard [29]. The test sample (10mg/ml) was heated in a boiling water bath for 10 min with 0.025M borax. Then 0.1% carbazole (in methanol) was added and boiling was continued for 15 min. The optical density was read at 540 nm.

Total sulfated polysaccharides were determined by metachromatic assay using heparin as standard [30]. 0.005% toluidine blue solution and 0.2% NaCl were added to the test sample (10mg/ml) and was mixed well for 30 sec. Then n-hexane was added to the above mixture and the 5ml aqueous layer was separated. Equal volume of absolute ethanol was added and the optical density was read at 631nm.

Fucose content was determined using cysteine hydrochloride [31]. Concentrated sulfuric acid was added to the test sample (10mg/ml) for 3 min. 3% cysteine hydrochloride was added and the difference in the optical density at 396nm and 427nm was calculated.

The monosaccharide xylose was estimated using orcinol method [32]. 0.1 ml of sample (10mg/ml) was heated in a boiling water bath for 30 min. Optical density was read at 670nm

In vitro antioxidant activity of isolated polysaccharides.

The antioxidant activity of all the three sulfated polysaccharides at different concentration (0.5mg/ml-2mg/ml) was determined by standard protocols. The antioxidant assays include DPPH (1-1-diphenyl 2-picryl hydrazyl) radical scavenging activity [33], hydroxyl radical scavenging activity [34], hydrogen peroxide Scavenging activity [35], Total Antioxidant Activity [36] and Reducing power [37].

Statistics

All the results were expressed as mean \pm standard deviation. One way ANOVA was calculated by using an online software statistic calculator. The P values <0.05 were considered to be significant [38], [39].

Results

The sulfated polysaccharides were obtained from *Sargassum swartzii*, *Ulva fasciata*

and *Chaetomorpha antennina* by ethanol precipitation and the total yield was found to be 11%, 1.5% and 1.3% respectively. The chemical compositions all the three algae are given in Table 1. *S. swartzii* showed a higher yield when compared to other two algae. It also showed an elevated amount of total carbohydrate, sulfate, uronic acid and fucose content when compared to other algae.

The antioxidant potential of crude sulfated polysaccharide from all the three algae was determined using various methods and compared with standard antioxidants. The DPPH radical scavenging activity is depicted in Figure 1A. All the samples exhibited a dose dependent increase in the DPPH scavenging activity. On comparing the scavenging potential of three algae, *S. swartzii* showed a higher DPPH scavenging effect. The half-maximal inhibitory concentration (IC₅₀) value for *S. swartzii* was 1.7mg/ml, for *U. fasciata* it was 3.8mg/ml and for *C. antennina* it was 6.2mg/ml. Ascorbic acid exhibited 50% scavenging ability at a concentration of 20 μ g/ml.

Figure 1B shows the hydroxyl radical scavenging activity of all the samples. Here also *S. swartzii* showed a better activity when compared to others. The half-maximal inhibitory concentration (IC₅₀) values for *S. swartzii*, *U. fasciata* and *C. antennina* was found to be 1.8, 2.6 and 1.9mg/ml respectively where as ascorbic acid control showed IC₅₀ at 282 μ g/ml. The scavenging ability of all samples was in a dose-dependent manner.

All the samples were potent dose-dependent inhibitors of hydrogen peroxide and exhibited a better effect below 0.5mg/ml (Figure 1C). At 0.5mg/ml, *S. swartzii* exhibited 71.2% scavenging of H₂O₂, for *U. fasciata* it was 62.7% and for *C. antennina* was 60.3%. Ascorbic acid showed 57% inhibition at a concentration of 20 μ g/ml.

The total antioxidant activity of sulfated polysaccharides from all the three algae is displayed in Figure 2A. Here also *S. swartzii* showed a good activity at higher concentration than at lower dose. 2mg/ml of *S. swartzii* showed 106.7 μ g/ml equivalence of ascorbic acid, whereas for *U. fasciata* and *C. antennina* it was 36.6 and 99.5 μ g/ml equivalence of ascorbic acid respectively. The positive control gallic acid (100 μ g/ml) was equivalent to 83 μ g/ml equivalence of ascorbic acid.

The reducing power of sulfated polysaccharide from *S. swartzii* was found to be more efficient

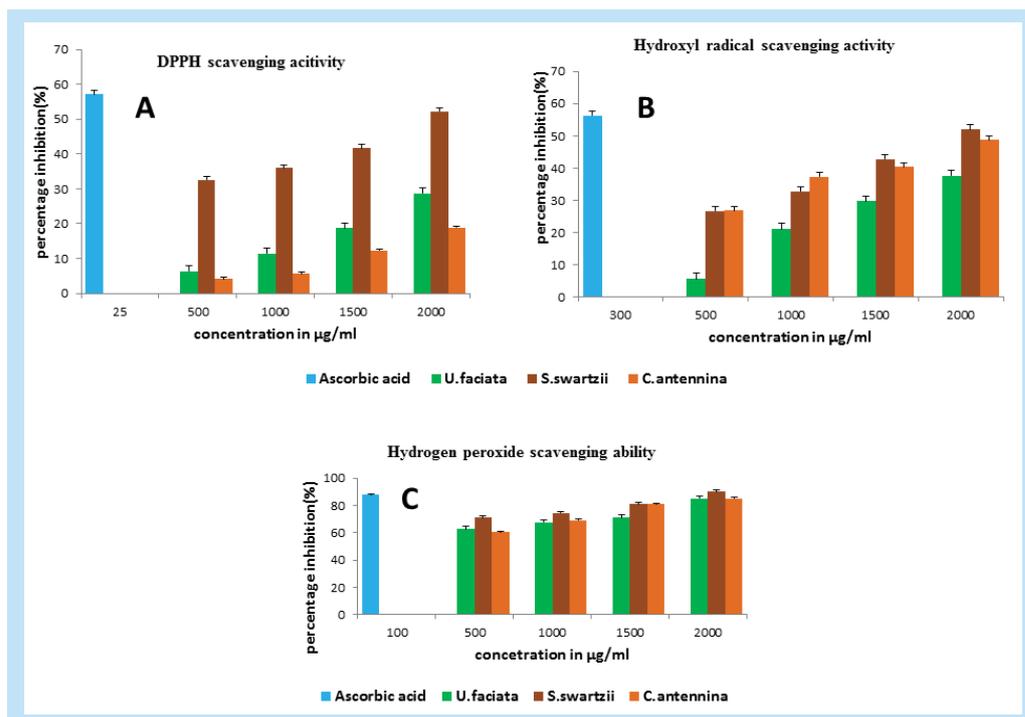


Figure 1: (A) DPPH radical scavenging activity of sulfated polysaccharides from *S. swartzii*, *U. fasciata* and *C. antennina* using ascorbic acid as the reference standard. The IC50 value for *S. swartzii* was 1.7mg/ml, for *U. fasciata* it was 3.8mg/ml and for *C. antennina* it was 6.2mg/ml. Ascorbic acid exhibited 50% scavenging ability at a concentration of 20µg/ml. (B) Hydroxyl radical scavenging potential of sulfated polysaccharides from *S. swartzii*, *U. fasciata* and *C. antennina*. Ascorbic acid was used as the positive control. The IC50 values for *S. swartzii*, *U. fasciata* and *C. antennina* was found to be 1.8, 2.6 and 1.9mg/ml respectively. (C) The hydrogen peroxide scavenging potential of sulfated polysaccharides from *S. swartzii*, *U. fasciata* and *C. antennina* using Ascorbic acid as reference standard. At 0.5mg/ml, *S. swartzii* exhibited 71.2% scavenging of H2O2. For *U. fasciata* it was 62.7% and for *C. antennina* 60.3% at the same concentration. Ascorbic acid showed 57% inhibition at a concentration of 20µg/ml.

Composition (%)	<i>Sargassum swartzii</i>	<i>Ulva fasciata</i>	<i>Chaetomorpha antennina</i>
Carbohydrates	12.9± 1.59	12.3 ±1.02	12.7 ±1.08
Sulfate	10.6± 0.57	2.3 ±0.13	1.5 ±0.11
Uronic acid	7.7 ±1.11	4 ±0.54	4.5 ±0.32
Fucose	2.9 ±0.22	2.2 ±0.05	3 ±0.16
Sulfated polysaccharide	1.2 ±0.19	0.15 ±0.12	2.2 ±0.91
Xylose	0.16 ±0.07	0.26 ±0.04	0.02 ±0.01

when compared to other two algae (Figure 2B). 2mg/ml of SPS from *S. swartzii*, *U. fasciata* and *C. antennina* showed 42, 22 and 24.4µg/ml equivalence of ascorbic acid respectively. 100µg/ml of gallic acid, the positive control, was equivalent to 85µg/ml equivalence of ascorbic acid.

Discussion

The study is aimed at the comparison of sulfated polysaccharides from three marine algae, *Sargassum swartzii*, *Ulva fasciata* and *Chaetomorpha antennina*, with respect to their chemical composition and antioxidant activity.

Acidic extraction method was applied for all the three algae. The yield and properties of isolated polysaccharides were highly influenced by the extraction procedure adopted for the same [40]. It is well known that the acidic extraction yields more sulfated polysaccharide than water extraction in *Sargassum* sps. [41]. Also in another method, which includes the precipitation of sulfated polysaccharide after removing alginate, the yield was found to be very less (4.3%) [42]. The brown algae *Sargassum swartzii* gave higher yield of sulfated polysaccharide when compared to other two green seaweeds. This higher yield is comparable to the yield obtained from *Sargassum*

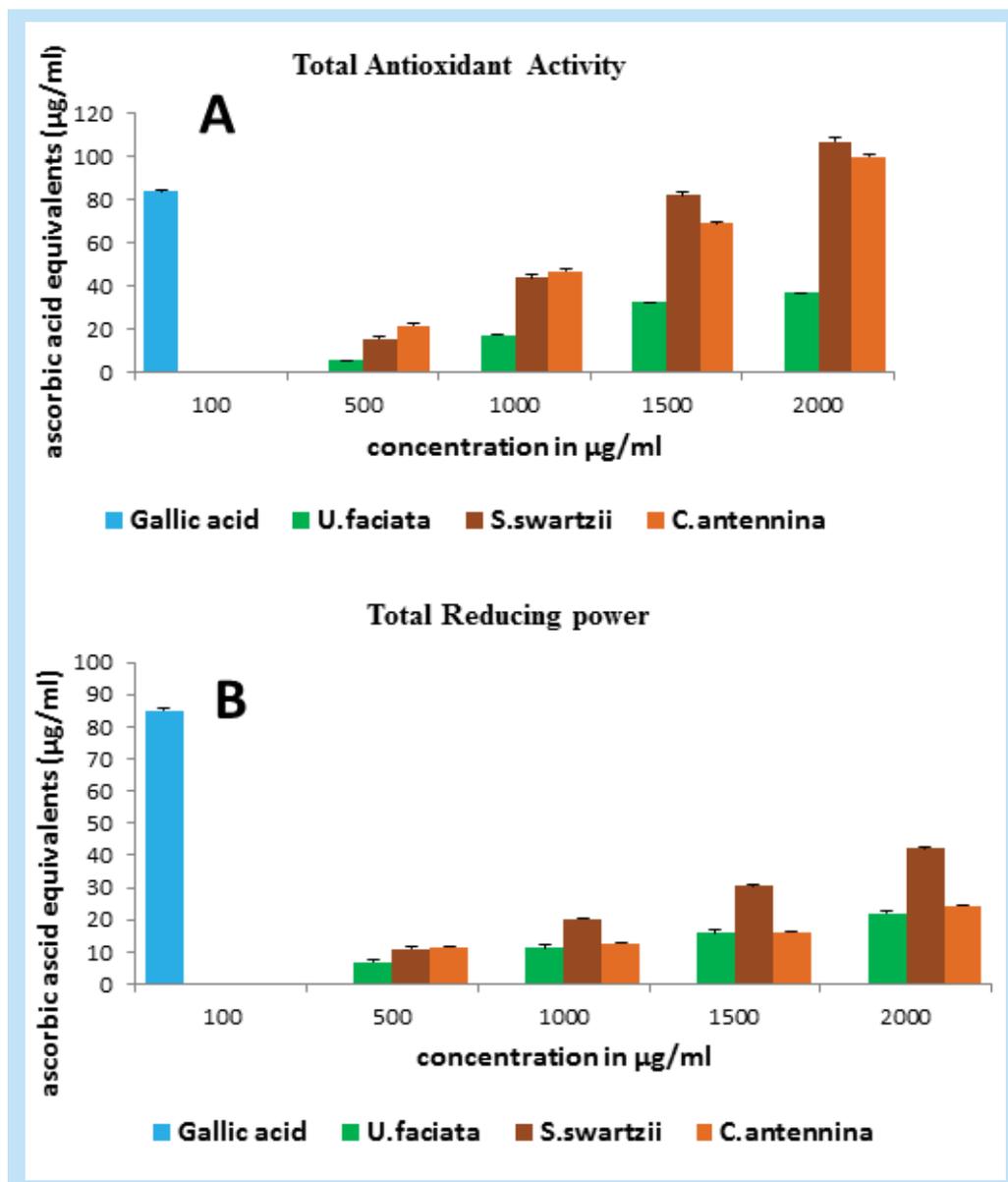


Figure 2: (A) The total antioxidant activity of sulfated polysaccharides from *S. swartzii*, *U. fasciata* and *C. antennina* using gallic acid as the positive control. 2mg/ml of *S. swartzii* showed 106.7µg/ml equivalence of ascorbic acid, whereas for *U. fasciata* and *C. antennina* it was 36.6 and 99.5µg/ml equivalence of ascorbic acid respectively. The positive control gallic acid (100µg/ml) was equivalent to 83µg/ml equivalence of ascorbic acid. (B) The reducing power of sulfated polysaccharides from *S. swartzii*, *U. fasciata* and *C. antennina*. 2mg/ml of SPS from *S. swartzii*, *U. fasciata* and *C. antennina* showed 42, 22 and 24.4µg/ml equivalence of ascorbic acid respectively. Gallic acid was used as the positive control.

polycystum, 11.32±0.12% (w/w) [41]. When the acidic extraction method was used for both the green algae in our study, the yield was very less. Extraction using hot water at 75–85°C, was found to be effective for polysaccharide isolation in *Ulva* species [43]. Thus, we could see that the current method is suitable for the isolation of sulfated polysaccharides from brown algae.

Sulfated polysaccharides are complex group of molecules and marine algae is its one of the major non-animal source [44]. They are composed of mainly sulfate and monosaccharides repeats

and display a wide structural diversity among different groups of marine algae [45]. Fucose, mannose, galactose, glucose, xylose and uronic acid are the main monosaccharides found in marine sulfated polysaccharides [18]. Earlier studies confirmed that the seaweeds contain large amount of polysaccharides [46]. All the three algal polysaccharides evaluated here were found to have almost similar and considerable percentage of total carbohydrate content. A sulfated polysaccharide isolated by Kokilam G et al. from *Sargassum tenerrimum* was found to

posses 8.21% of total carbohydrate [47]. Also in another report, the biochemical analysis of *Chaetomorpha antennina* [48] exhibited 19.68% and 18.4% of carbohydrate content respectively. Total carbohydrate content of the current three algae also correlates with these results.

The brown algae *Sargassum swartzii* possess very high amount of sulfate content (10.6%) when compared to other two green algae. Evaluation of sulfate content of another brown algae *Sargassum polycystum* also reported comparable fraction of sulfate [41]. In another study, it was reported that brown algae possess more sulfate content than green algae [49]. Fucose is the most important monosaccharide in brown algae and our samples contained significant amount of fucose. But there is also presence of fucose in green algal polysaccharide. The fucose can also be present in the polysaccharides of green algae in rare cases [50]. Uronic acid content was found to be high in *Sargassum* than the other two green algae and similar trend is already reported [45] Brown algae are known to contain high amount of uronic acid in their polysaccharide structure [21].

The antioxidant capacity of sulfated polysaccharides from seaweeds is well-studied [51]. They are known to possess various antioxidant activities such as scavenging of free radicals like superoxide, hydroxyl and DPPH, lipid peroxide inhibition and ferric reducing antioxidant power [52]. In the current study the antioxidant property of all the three sulfated polysaccharides were evaluated and all of them exhibited antioxidant effects in a concentration dependent manner.

DPPH radical scavenging assay is a simple method for the determination of antioxidant capacity of a compound. DPPH (2,2-diphenyl-1-picryl-hydrazyl-hydrate) is a stable free radical that gives purple color in ethanol solution and on reduction in the presence hydrogen donating antioxidants, turns the solution colorless [53]. Thus, the characteristic absorption shown by DPPH reduce in accordance with increased concentration of the antioxidant compound, which indicates the DPPH scavenging potential of the compound [54]. In the present examination, sulfated polysaccharides from all the three algae exhibited a dose dependent increase in the scavenging of DPPH radical and *Sargassum* was found to be more efficient. However, the effective concentration was found to be 2mg or above in all the three cases, which is much higher when compared to ascorbic acid.

There are reports that substantiate this high concentration of sulfated polysaccharide from seaweeds for DPPH scavenging [54, 55].

The hydroxyl radical scavenging effect of the samples is determined by using Fenton reaction. Here, hydroxyl radicals are generated using Ferric-ascorbate–EDTA–H₂O₂ system, which will later react with deoxyribose to produce thiobarbituric acid reactive substances (TBARS). When heated with Thiobarbituric acid (TBA), TBARS forms a pink chromogen and the intensity of the color will get reduced depending up on the antioxidant ability of the test compound [34]. Evaluation of the current samples results in the effective scavenging of hydroxyl radical in a dose dependent manner. At a concentration of 2mg/ml sulfated polysaccharides from *Sargassum swartzii* was found to be more effective in scavenging hydroxyl radical. In many literatures, it was reported that sulfated polysaccharides exhibit moderate or no defense against hydroxyl radical [56].

Hydrogen peroxide is a weak oxidizing agent that can cross membranes very rapidly and can oxidize and inactivate many essential enzymes. Thus, the cells have to eliminate H₂O₂ immediately for their biological existence [57]. The H₂O₂ scavenging ability of an antioxidant compound can be evaluated by means of decrease in the optical density. Here sulfated polysaccharides from all the three algae showed a better ability to scavenge H₂O₂ in a dose dependent manner. All the three samples showed a better activity at a concentration below 0.5mg/ml. This is much better than other reports on the H₂O₂ scavenging ability of sulfated polysaccharides [58]. The sulfated polysaccharide isolated from the red algae *Pterocladia capillacea* showed 45.76% inhibition of hydrogen peroxide at a concentration of 1mg/ml [59].

The total antioxidant activity of the samples is evaluated as their ability to reduce molybdenum VI to molybdenum V and form the green colored phosphomolybdenum complex. There will be an increase in the intensity of the green color as the concentration of the sample increases [36]. Even though sulfated polysaccharides from *Sargassum swartzii* showed more activity, it is much lower when compared to that of standard gallic acid. Reducing power of a compound is another potent indicator of its antioxidant activity. It is the ability of the compound to reduce ferric ion to ferrous form, which is associated with a color change from yellow to Pearl's Prussian Blue [60]. In the present study, all the samples exhibited a

dose dependent increase in the reducing power. But, when compared to standard gallic acid it is too low. Thus total antioxidant and reducing power contributes much less to the antioxidant capacity of the sulfated polysaccharides in our study.

In all the antioxidant assays, sulfated polysaccharides from the brown algae *Sargassum swartzii* showed better activity when compared to other two algae. This can be explained easily by the percentage of sulfate in the extracted polysaccharides. It has already proven that, sulfate content is the main factor that contributes to the biological activity of the sulfated polysaccharides [51]. This may vary with degree of sulfation and position of sulfate groups [61]. Since the polysaccharide isolated from *Sargassum swartzii* contains higher amount of sulfate, it can scavenge free radicals more effectively. The possible mechanism behind the ROS scavenging potentials of sulfated polysaccharide was suggested to be H-atom transfer and electron transfer between polysaccharides and free radicals [62].

Conclusion

In conclusion, the sulfated polysaccharides isolated from all the three marine algae contained significant amount of carbohydrate. In brown algae *Sargassum swartzii*, sulfate content was found to be higher than the other two. All samples exhibited potent antioxidant capacity and the higher activity of *Sargassum swartzii* can be due to the presence high sulfate. Thus, these SPS can be effectively used for scavenging ROS *in vitro* and thus can reduce the risk of many diseases. Potential application of marine derived, novel antioxidant, sulfated polysaccharides in the food industry reveals a wide scope of these edible seaweeds.

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